



Material Safety Data Sheet

Iodine MSDS

Section 1: Chemical Product

Product Name: Iodine

Catalog Codes: SLI1513

CAS#: 7553-56-2

RTECS: NN1575000

TSCA: TSCA 8(b) inventory: Iodine

CI#: Not available.

Synonym:

Chemical Name: Iodine

Chemical Formula: I₂

Section 2: Composition and Information on Ingredients

Composition:

Name CAS # % by Weight

Iodine 7553-56-2 100

Toxicological Data on Ingredients: Iodine: ORAL (LD50): Acute: 14000 mg/kg [Rat]. 22000 mg/kg [Mouse].

Section 3: Hazards Identification

Potential Acute Health Effects:

Very hazardous in case of skin contact (irritant), of eye contact (irritant), of ingestion, of inhalation. Hazardous in case of skin contact (corrosive), of eye contact (corrosive). Slightly hazardous in case of skin contact (permeator). The amount of tissue damage depends on length of contact. Eye contact can result in corneal damage or blindness. Skin contact can produce inflammation and blistering. Inhalation of dust will produce irritation to gastro-intestinal or respiratory tract, characterized by burning, sneezing and coughing. Severe over-exposure can produce lung damage, choking, unconsciousness or death. Inflammation of the eye is characterized by redness, watering, and itching. Skin inflammation is characterized by itching, scaling, reddening, or, occasionally, blistering.

Potential Chronic Health Effects:

Hazardous in case of skin contact (irritant), of eye contact (irritant), of ingestion, of inhalation.

CARCINOGENIC EFFECTS: Not available.

MUTAGENIC EFFECTS: Not available.

TERATOGENIC EFFECTS: Not available.

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DEVELOPMENTAL TOXICITY: Not available.

The substance is toxic to thyroid.

The substance may be toxic to blood, kidneys, liver, skin, eyes.

Repeated or prolonged exposure to the substance can produce target organs damage. Repeated exposure of the

eyes to a low level of dust can produce eye irritation. Repeated skin exposure can produce local skin destruction,

or dermatitis. Repeated inhalation of dust can produce varying degree of respiratory irritation or lung damage.

Section 4: First Aid Measures

Eye Contact:

Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at

least 15 minutes. Cold water may be used. WARM water MUST be used. Get medical attention immediately.

Skin Contact:

In case of contact, immediately flush skin with plenty of water for at least 15 minutes while removing contaminated

clothing and shoes. Cover the irritated skin with an emollient. Wash clothing before reuse.

Thoroughly clean

shoes before reuse. Get medical attention immediately.

Serious Skin Contact:

Wash with a disinfectant soap and cover the contaminated skin with an anti-bacterial cream.

Seek medical

attention.

Inhalation:

If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get

medical attention.

Serious Inhalation:

Evacuate the victim to a safe area as soon as possible. Loosen tight clothing such as a collar, tie, belt or

waistband. If breathing is difficult, administer oxygen. If the victim is not breathing, perform mouth-to-mouth

resuscitation. **WARNING:** It may be hazardous to the person providing aid to give mouth-to-mouth resuscitation

when the inhaled material is toxic, infectious or corrosive. Seek immediate medical attention.

Ingestion:

Do NOT induce vomiting unless directed to do so by medical personnel. Never give anything by mouth to an

unconscious person. Loosen tight clothing such as a collar, tie, belt or waistband. Get medical attention if

symptoms appear.

Serious Ingestion: Not available.

Section 5: Fire and Explosion Data

Flammability of the Product: Non-flammable.

Auto-Ignition Temperature: Not applicable.

Flash Points: Not applicable.

Flammable Limits: Not applicable.

Products of Combustion: Not available.

Fire Hazards in Presence of Various Substances: Not applicable.

Explosion Hazards in Presence of Various Substances:

Risks of explosion of the product in presence of mechanical impact: Not available.

Risks of explosion of the product in presence of static discharge: Not available.

Fire Fighting Media and Instructions: Not applicable.

Special Remarks on Fire Hazards:

Ignition on contact with bromine, ... chlorine trifluoride, ...metals (powdered) + water, aluminum-titanium alloys + heat, metal acetylides, ... nonmetals, ... sodium phosphinate. Incandescent reaction with cesium oxide (above 150 deg C), bromine trifluoride, metal acetylides or carbides [e.g. barium acetylide (above 122 deg C), calcium acetylide (above 305 deg C), strontium acetylide (above 182 deg C), zirconium acetylide (above 400 degC)]. Magnesium burns vigorously when heated with iodine vapor. Iodine unites with fluorine at ordinary temperature with a luminous flame

Special Remarks on Explosion Hazards:

Explosive reactions with iodine and: hafnium powder + heat; tetraamine copper (II) sulfate + ethanol; trioxygen difluoride; polyacetylene (at 113 deg. C); potassium; sodium; butadiene+ ethanol +mercuric oxide;

Section 6: Accidental Release Measures

Small Spill: Use appropriate tools to put the spilled solid in a convenient waste disposal container.

Large Spill:

Corrosive solid.

Stop leak if without risk. Do not get water inside container. Do not touch spilled material. Use water spray to reduce vapors. Prevent entry into sewers, basements or confined areas; dike if needed. Call for assistance on disposal. Be careful that the product is not present at a concentration level above TLV. Check TLV on the MSDS and with local authorities.

Section 7: Handling and Storage

Precautions:

Keep container dry. Do not ingest. Do not breathe dust. Never add water to this product. In case of insufficient ventilation, wear suitable respiratory equipment. If ingested, seek medical advice immediately and show the container or the label. Avoid contact with skin and eyes. Keep away from incompatibles such as oxidizing agents, reducing agents, metals.

Storage: Keep container tightly closed. Keep container in a cool, well-ventilated area. Do not store above 25°C (77°F).

Section 8: Exposure Controls/Personal Protection

Engineering Controls:

Use process enclosures, local exhaust ventilation, or other engineering controls to keep airborne levels below recommended exposure limits. If user operations generate dust, fume or mist, use ventilation to keep exposure to airborne contaminants below the exposure limit.

Personal Protection:

Splash goggles. Synthetic apron. Vapor and dust respirator. Be sure to use an approved/certified respirator or equivalent. Gloves.

Personal Protection in Case of a Large Spill:

Splash goggles. Full suit. Vapor and dust respirator. Boots. Gloves. A self contained breathing apparatus should be used to avoid inhalation of the product. Suggested protective clothing might not be sufficient; consult a

specialist BEFORE handling this product.

Exposure Limits:

STEL: 1 (mg/m³) from ACGIH (TLV) [United States]

STEL: 0.1 (ppm) from ACGIH (TLV) [United States]

TWA: 1 CEIL: 1 (mg/m³) from OSHA (PEL) [United States]

TWA: 0.1 CEIL: 0.1 (ppm) from OSHA (PEL) [United States]

STEL: 0.1 (ppm) [United Kingdom (UK)]

STEL: 1.1 (mg/m³) [United Kingdom (UK)] Consult local authorities for acceptable exposure limits.

Section 9: Physical and Chemical Properties

Physical state and appearance: Solid.

Odor: Sharp Characteristic. (Strong.)

Taste: Not available.

Molecular Weight: 253.81 g/mole

Color: Purple solid with metallic luster. (Dark.)

pH (1% soln/water): Not available.

Boiling Point: 184.4 °C (363.9 °F)

Melting Point: 113.7 °C (236.7 °F)

Critical Temperature: Not available.

Specific Gravity: 4.93 (Water = 1)

Vapor Pressure: Not applicable.

Vapor Density: Not available.

Volatility: Not available.

Odor Threshold: Not available.

Water/Oil Dist. Coeff.: The product is more soluble in oil; log(oil/water) = 2.5

Ionicity (in Water): Not available.

Dispersion Properties: See solubility in water, methanol, diethyl ether.

Solubility:

Easily soluble in diethyl ether. Soluble in methanol.

Very slightly soluble in cold water, hot water.

Section 10: Stability and Reactivity Data

Stability: The product is stable.

Instability Temperature: Not available.

Conditions of Instability: Heat, direct sunlight, incompatible materials

Incompatibility with various substances: Reactive with oxidizing agents, reducing agents, metals.

Corrosivity:

Extremely corrosive in presence of steel, of stainless steel(304), of stainless steel(316).

Non-corrosive in presence of glass, of aluminum, of copper.

Special Remarks on Reactivity:

Incompatible with liquid chlorine, acetaldehyde, ammonia, salt + ethanol, ammonium hydroxide, methyl alcohol,

antimony, silver azide, lithium, potassium, sodium, phosphorous, bromine pentafluoride, fluorine, oxygen

difluoride, magnesium, finely divided metals, organic solvents, rubber goods, plastics, zinc, aluminum, alkali

metals, sulphur, ammonia solutions, Bromine trifluoride, reducing agents, iron, ethanol + butadiene; ethanol +

phosphorous; ethanol + methanol + HgO; foramide + pyridine + sulfur trioxide; formamide; halogens or

interhalogens; mercuric oxide; metal carbides; oxygen; pyridine; sodium hydride.

Violent reaction with iodine and aluminum + diethyl ether ... (and) titanium (above 113 deg C)

Special Remarks on Corrosivity:

Corrodes steel. No corrosive effect on bronze

Polymerization: Will not occur.

Section 11: Toxicological Information

Routes of Entry: Inhalation. Ingestion.

Toxicity to Animals: Acute oral toxicity (LD50): 14000 mg/kg [Rat].

Chronic Effects on Humans:

Causes damage to the following organs: thyroid.

May cause damage to the following organs: blood, kidneys, liver, skin, eyes.

Other Toxic Effects on Humans:

Very hazardous in case of skin contact (irritant), of ingestion, of inhalation.

Hazardous in case of skin contact (corrosive), of eye contact (corrosive).

Slightly hazardous in case of skin contact (permeator).

Special Remarks on Toxicity to Animals:

Lowest Published Lethal Dose:

LDL [Human] - Route: Oral; Dose: 28 mg/kg

LCL [Rat] - Route: Inhalation; Dose: 137 ppm/1H

Special Remarks on Chronic Effects on Humans: May cause adverse reproductive effects (effects on new born).

Special Remarks on other Toxic Effects on Humans:

Acute Potential Health Effects:

Skin: Corrosive action skin. Causes skin irritation and burns. It is corrosive and can cause penetrating lesions and brown staining. It can be absorbed by the skin.

Eyes: Causes eye irritation and burns. May cause conjunctivitis. Exposure to vapor can cause burning

sensation in the eyes, tearing, inflammation of the eye lids. Exposure to high concentrations of vapor can cause

Dendritic Keratitis in which the corneal epithelium is sloughed off.

Inhalation: Excessive inhalation of iodine vapors may cause respiratory tract, nasal, and mucous membrane

irritation with possible burns. Symptoms may include coughing, tightness in the chest, burning sensations to the

mucosal, tracheal, and pulmonary tissues, rhinitis, dyspnea/respiratory distress, coughing, sneezing, pulmonary

edema, chemical pneumonitis, edema of the larynx and bronchi, pharyngitis, swelling of the parotid gland, and

cachexia. High exposure may lead to lung disease and may also affect behavior/central nervous system (delirium,

hallucination, depression, seizure, dizziness, headache, stupor, somnolence).

Ingestion: Ingestion of large doses may cause irritation of mouth of the digestive tract with thirst, nausea,

vomiting, abdominal pain, hypermotility, and diarrhea, staining of mouth, esophagus, lips, mucous membranes,

metallic taste, abdominal pain, fever. It may also affect the cardiovascular system (tachycardia, hypotension,

cardiovascular collapse), behavior/central nervous system (delirium, dizziness, headache, hallucinations, seizures,

Section 12: Ecological Information

Ecotoxicity: Not available.

BOD5 and COD: Not available.

Products of Biodegradation:

Possibly hazardous short term degradation products are not likely. However, long term degradation products may arise.

Toxicity of the Products of Biodegradation: The product itself and its products of degradation are not toxic.

Special Remarks on the Products of Biodegradation: Not available.

Section 13: Disposal Considerations

Waste Disposal:

Waste must be disposed of in accordance with federal, state and local environmental

control regulations.

Section 14: Transport Information

DOT Classification: Class 8: Corrosive material

Identification: : Corrosive Solid, n.o.s (Iodine) UNNA: 1759 PG: III

Special Provisions for Transport: Not available.

Section 15: Other Regulatory Information

Federal and State Regulations:

Illinois toxic substances disclosure to employee act: Iodine Rhode Island RTK hazardous substances: Iodine

Pennsylvania RTK: Iodine Minnesota: Iodine

Massachusetts RTK: Iodine Massachusetts spill list: Iodine

New Jersey: Iodine

California Director's List of Hazardous Substances: Iodine

TSCA 8(b) inventory: Iodine

Other Regulations:

OSHA: Hazardous by definition of Hazard Communication Standard (29 CFR 1910.1200).

EINECS: This product is on the European Inventory of Existing Commercial Chemical Substances.

Other Classifications:

WHMIS (Canada):

CLASS D-2A: Material causing other toxic effects (VERY TOXIC). CLASS E: Corrosive solid.

DSCL (EEC):

R38- Irritating to skin. R41- Risk of serious damage to eyes.

S2- Keep out of the reach of children. S26- In case of contact with eyes, rinse immediately with plenty of water and seek medical advice.

S39- Wear eye/face protection. S46- If swallowed, seek medical advice immediately and show this container or label.

HMIS (U.S.A.):

Health Hazard: 3

Fire Hazard: 0

Reactivity: 0

Personal Protection: j

National Fire Protection Association (U.S.A.):

Health: 3

Flammability: 0

Reactivity: 0

Specific hazard:

Protective Equipment:

Gloves.

Synthetic apron.

Vapor and dust respirator. Be sure to use an approved/certified respirator or equivalent. Wear appropriate respirator when ventilation is inadequate.

Splash goggles.

Section 16: Other Information

References: Not available.

Other Special Considerations: Not available.

<http://www.sparchem.com/>